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*Sharafat H. Mammadov^a, Guseyn R. Gurbanov^b, Rana A. Ismailova^b***PHASE DIAGRAM OF THE AgGaS_2 – PbGa_2S_4 SYSTEM**^a Institute of Catalysis and Inorganic Chemistry named after academician M. Naghiyev, Baku, Republic of Azerbaijan^b Azerbaijan State Oil and Industry University, Baku, Republic of Azerbaijan

Lead and silver thiogallates are notable for their practical applications in nonlinear optical devices, detectors, solar cells, photodiodes, and phosphors. This study aims to develop materials with multifunctional properties and investigate the AgGaS_2 – PbGa_2S_4 system. The initial sulfides, AgGaS_2 and PbGa_2S_4 , were synthesized from high-purity elements sealed in evacuated quartz ampoules (0.133 Pa). Complex alloys of the AgGaS_2 – PbGa_2S_4 system were prepared at 1200–1350 K. Polycrystalline samples were annealed at 850 K for 270 hours. Phase equilibria in the AgGaS_2 – PbGa_2S_4 system were analyzed using a combination of physicochemical methods, including differential thermal analysis, powder X-ray diffraction, microstructural analysis, microhardness, and density measurements, and the phase diagram was constructed. The system was identified as a quasi-binary section of the quasi-ternary Ag_2S – Ga_2S_3 – PbS system, belonging to the eutectic type. The eutectic point was determined at 1100 K and 55 mol.% PbGa_2S_4 . Solid solution regions were identified based on the primary components. At room temperature, the AgGaS_2 solid solution region extends to 10 mol.% PbGa_2S_4 , while the PbGa_2S_4 solid solution region extends to 18 mol.% AgGaS_2 . At the eutectic temperature, these solubilities increase to 20 and 25 mol.%, respectively. AgGaS_2 -based solid solutions crystallize in chalcopyrite-type structures, while PbGa_2S_4 -based solid solutions adopt orthorhombic structures.

Keywords: PbS – Ga_2S_3 – Ag_2S , quasi-binary system, solid solutions, AgGaS_2 , ternary system, eutectic.

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Introduction

Crystals of silver and lead thiogallates have recently aroused particular interest in the study of systems containing sulfides with the formulae $\text{A}^{\text{I}}\text{B}^{\text{III}}\text{C}_2^{\text{VI}}$ and AB_2C_4 . First of all, this is due to the emerging possibilities of their practical use in the manufacture of nonlinear optical devices, detectors, solar cells, photodiodes and phosphors [1–9]. The study of the optical properties of PbGa_2S_4 lead thiogallate crystals revealed that it is a promising material for optoelectronics [10].

In the PbS – Ga_2S_3 system [11,12], compounds of PbGa_2S_4 and $\text{Pb}_2\text{Ga}_2\text{S}_5$ are formed. According to ref. [11], both compounds are formed by a peritectic

reaction and melt with decomposition at temperatures of 1203 and 1173 K, respectively. In contrast to ref. [11], study [12] found that PbGa_2S_4 and $\text{Pb}_2\text{Ga}_2\text{S}_5$ melt congruently at 1163 and 1033 K, respectively; and in general, the PbS – Ga_2S_3 system behaves like the Cu_2S – Ga_2S_3 and SrS – Ga_2S_3 systems. According to ref. [13], PbGa_2S_4 is a phase of variable composition. The PbGa_2S_4 compound belongs to the EuGa_2S_4 structural type and crystallizes in the orthorhombic system ($a=20.44$ Å, $b=20.64$ Å, $c=12.09$ Å; space group Fddd , $z=32$, $\rho=4.94$ g/cm³ [11]).

Study of the Ag_2S – Ga_2S_3 system showed that three compounds are formed in the system with the compositions AgGaS_2 , Ag_9GaS_6 and $\text{Ag}_2\text{Ga}_{20}\text{S}_{31}$.

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Sharafat H. Mammadov, Guseyn R. Gurbanov, Rana A. Ismailova

$\text{Ag}_2\text{Ga}_{20}\text{S}_{31}$ is formed by a peritectic reaction at a temperature of 1263 K, and the compounds AgGaS_2 and Ag_9GaS_6 are congruent at 1270 K and 1063 K, respectively. The compounds AgGaS_2 and Ag_9GaS_6 crystallize in the chalcopyrite structural type: $a=5.7544 \text{ \AA}$, $c=10.299 \text{ \AA}$; space group I42d. The Ag_9GaS_6 compound crystallizes into orthorhombic systems with parameters $a=10.777 \text{ \AA}$, $b=7.706 \text{ \AA}$, $c=7.605 \text{ \AA}$.

The purpose of this research work was to obtain materials with multifunctional properties and study the AgGaS_2 – PbGa_2S_4 system.

Experimental

The alloys for studying the AgGaS_2 – PbGa_2S_4 system were prepared from master alloys in evacuated quartz ampoules at a temperature of 1200–1350 K. Pure elements were used as starting materials (Ag, highly pure grade; Ga, G1-000 grade; sulfur highly pure grade; and Pb containing main substance no less than 99.999%).

Polycrystalline samples were annealed at 850 K for 270 hours. The pressed alloys of the AgGaS_2 – PbGa_2S_4 system were studied by differential thermal analysis (DTA), powder X-ray diffraction (XRD), microstructural analysis (MSA microscope MIM-7), measuring the density and microhardness (PMT-3 device). DTA was performed using NETZSCH 404 F1 Pegasus at room temperature to approximately 1400 K, depending on the alloy composition, at a heating rate of 10 K/min. Temperature accuracy was $\pm 2 \text{ K}$. Powder X-ray diffraction data were examined on a Bruker D2 Phaser diffractometer using $\text{CuK}\alpha 1$ radiation ($\lambda=1.54056 \text{ \AA}$).

Results and discussions

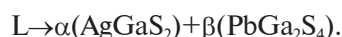
In order to study the nature of the interaction in the AgGaS_2 – PbGa_2S_4 system, 12 alloys were synthesized (Table).

The DTA results revealed two endothermic effects in the thermograms of alloys containing 0–100% PbGa_2S_4 . Microstructural analysis data showed that alloys similar in composition to the ternary compounds AgGaS_2 and PbGa_2S_4 (100–90 and 100–82 mol.%, respectively) consist of one phase; with an increase in the content of the second component, two-phase regions are noted. In the subsolidus of the system, in the concentration range of 10–82 mol.% PbGa_2S_4 , two phases ($\alpha+\beta$) co-crystallize. The XRD results of alloys of the AgGaS_2 – PbGa_2S_4 system are consistent with the MSA and DTA data and confirm the existence of solid solutions based on AgGaS_2 and PbGa_2S_4 . Considering the results of DTA, MSA, XRF and microhardness measurements, a state diagram of the AgGaS_2 – PbGa_2S_4 system was constructed (Fig. 1).

Results of thermal and microstructural analysis of alloys in the AgGaS_2 – PbGa_2S_4 system

Composition, mol.%		Thermal effects, K	Phase composition
AgGaS_2	PbGa_2S_4		
100	0	1270	α
90	10	1180, 1245	α
80	20	1115, 1205	$\alpha+\beta$
70	30	1100, 1175	$\alpha+\beta$
60	40	1100, 1145	$\alpha+\beta$
50	50	1100, 1125	$\alpha+\beta$
45	55	1100 (eutectic)	$\alpha+\beta$
40	60	110, 1130	$\alpha+\beta$
30	70	1105, 1145	$\alpha+\beta$
18	82	1110, 1155	β
10	90	1145, 1160	β
0	100	1165	β

As can be seen from Fig. 1, the system is a quasi-binary section of the quasi-ternary system Ag_2S – Ga_2S_3 – PbS and belongs to the eutectic type. The liquidus system consists of two branches of the primary crystallization phase AgGaS_2 and PbGa_2S_4 intersecting the eutectic point, characterizing non-variant equilibrium



The coordinates of the eutectic point are as follows: 55 mol.% PbGa_2S_4 and $T=1100 \text{ K}$. Based

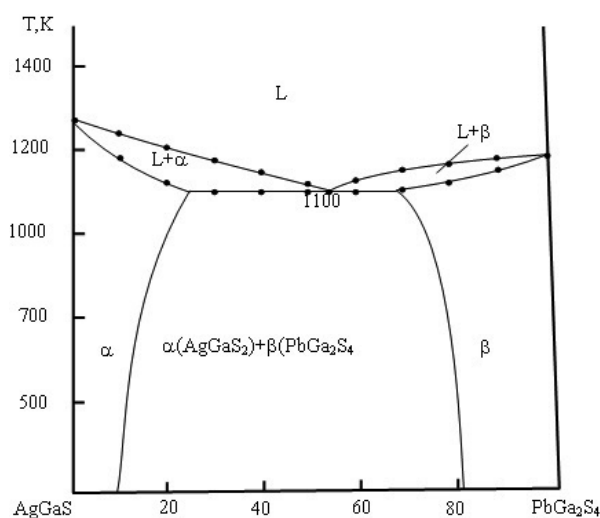


Fig. 1. Phase diagram of the AgGaS_2 – PbGa_2S_4 system

on the initial components, limited areas of solid solutions are formed. The results of X-ray phase analysis showed that the diffraction patterns of alloys containing 0–10 mol.% PbGa_2S_4 and 0–18 mol.% AgGaS_2 are identical to the diffraction patterns of the parent ternary compounds AgGaS_2 and PbGa_2S_4 , respectively. They are substitutional solid solutions. Based on the data obtained, the microhardness of the alloys of the AgGaS_2 – PbGa_2S_4 system was measured and a microhardness (H_μ) vs. composition diagram was constructed (Fig. 2).

The graph shows that the microhardness values of samples rich in AgGaS_2 and PbGa_2S_4 increase significantly from 3860 to 3990 MPa and from 2250 to 2400 MPa with the formation of solid solutions.

The densities of samples belonging to the

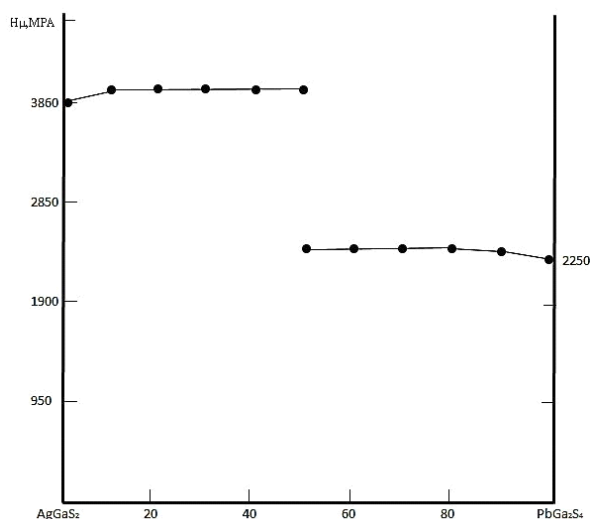


Fig. 2. H_μ vs. x diagrams of the AgGaS_2 – PbGa_2S_4 system

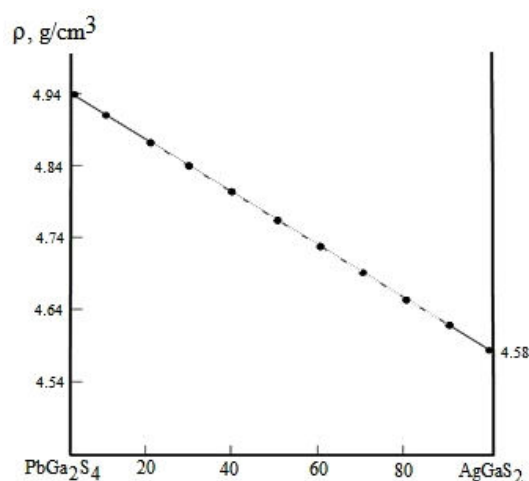


Fig. 3. ρ vs. x Diagram of the PbGa_2S_4 – AgGaS_2 system

AgGaS_2 – PbGa_2S_4 system were measured and a density vs. composition diagram was constructed (Fig. 3). It showed that the density of the samples varies within the density of the original sulfides (AgGaS_2 and PbGa_2S_4).

Conclusions

1. Using physical and chemical analysis methods (DTA, XRF, and MSA), the T vs. x phase diagram of the AgGaS_2 – PbGa_2S_4 system was constructed. It was determined that the AgGaS_2 – PbGa_2S_4 system is a quasi-binary cross-section of the quasi-ternary system Ag_2S – Ga_2S_3 – PbS and belongs to the eutectic type.

2. In the AgGaS_2 – PbGa_2S_4 system, the formation of solid solutions of the initial components was revealed. The solubility based on AgGaS_2 at room temperature (300 K) is 10 mol.% PbGa_2S_4 , and that on the basis of PbGa_2S_4 is 18 mol.% AgGaS_2 .

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ФАЗОВА ДІАГРАМА СИСТЕМИ AgGaS₂–PbGa₂S₄

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Тіогаллати свинцю та срібла вирізняються широкими можливостями практичного застосування в нелінійних оптичних пристроях, детекторах, сонячних батареях, фотодіодах та люмінофорах. Метою цього дослідження є розробка матеріалів із багатофункціональними властивостями та вивчення системи AgGaS₂–PbGa₂S₄. Вихідні сульфід AgGaS₂ і PbGa₂S₄ були синтезовані з високо-чистих елементів у кварцових ампулах, евакуйованих до тиску 0,133 Па. Комплексні сплави системи AgGaS₂–PbGa₂S₄ були одержані за температури 1200–1350 К. Полікристалічні зразки відпалювали за температури 850 К протягом 270 годин. Фазові рівноваги в системі AgGaS₂–PbGa₂S₄ були досліджені за допомогою комплексу фізико-хімічних методів, включаючи диференціальний термічний аналіз, рентгенівську порошкову дифракцію, мікρο-структурний аналіз, вимірювання мікротвердості та густини, що дозволило побудувати фазову діаграму. Було встановлено, що система AgGaS₂–PbGa₂S₄ є квазібінарним перерізом квазі-потрійної системи Ag₂S–Ga₂S₃–PbS і належить до евтектичного типу. Евтектична точка має координати: температура 1100 К і вміст 55 мол.% PbGa₂S₄. Ділянки твердих розчинів ідентифіковано на основі початкових компонентів. За кімнатної температури ділянка твердих розчинів на основі AgGaS₂ розширюється до 10 мол.% PbGa₂S₄, а на основі PbGa₂S₄ – до 18 мол.% AgGaS₂. За евтектичної температури ці розчинності зростають до 20 і 25 мол.%, відповідно. Тверді розчини на основі AgGaS₂ кристалізуються в структурах типу халькопіриту, тоді як тверді розчини на основі PbGa₂S₄ мають орторомбічну структуру.

Ключові слова: PbS–Ga₂S₃–Ag₂S; квазібінарна система; тверді розчини; AgGaS₂; потрійна система; евтектика.

PHASE DIAGRAM OF THE AGGaS₂–PBGa₂S₄ SYSTEM

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Keywords: PbS–Ga₂S₃–Ag₂S; quasi-binary system; solid solutions; AgGaS₂; ternary system; eutectic.

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